TEACHER GUIDE

9th-12th Grade

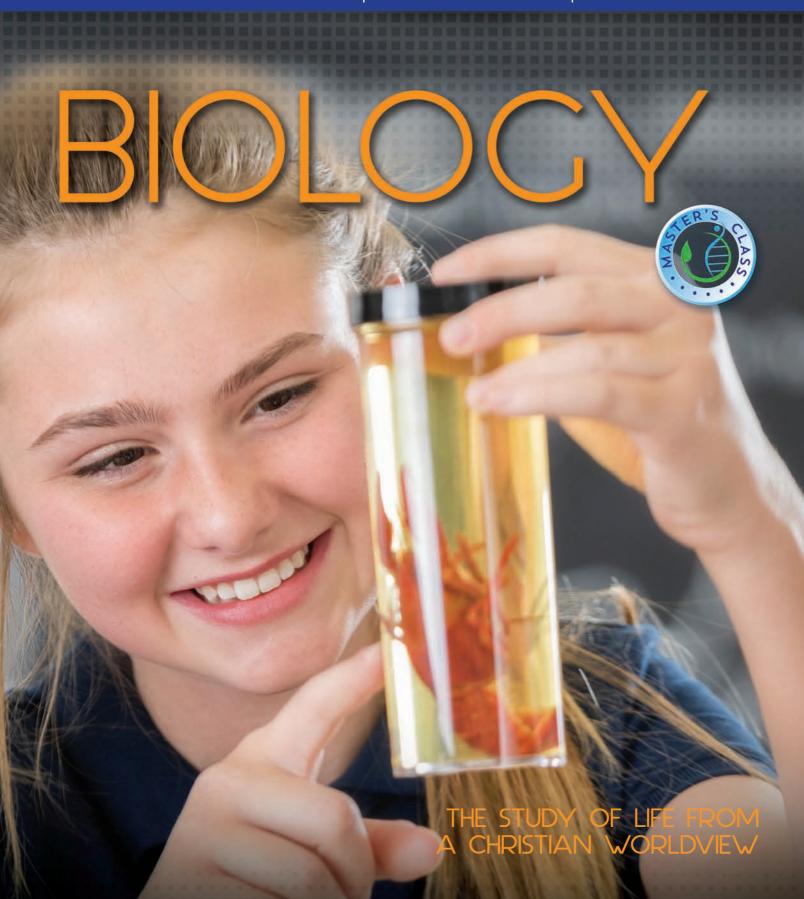
Includes Student Worksheets

Science

Weekly Lesson Schedule Labs







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9th-12th Grade

Includes Student Worksheets

Science



Master's Class Biology



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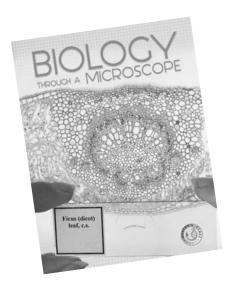
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Optional lab book available.



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B.A., Westmont College M.S., California State University, Northridge Ed.D., University of Southern California

Using This Teacher Guide

Features: The suggested weekly schedule enclosed has easy-to-manage lessons that guide the reading, worksheets, and all assessments. The pages of this guide are perforated and three-hole punched so materials are easy to tear out, hand out, grade, and store. Teachers are encouraged to adjust the schedule and materials needed in order to best work within their unique educational program.

Lesson Scheduling: Students are instructed to read the pages in their book and then complete the corresponding section provided by the teacher. Assessments that may include worksheets, activities, quizzes, and tests are given at regular intervals with space to record each grade. Space is provided on the weekly schedule for assignment dates, and flexibility in scheduling is encouraged. Teachers may adapt the scheduled days per each unique student situation. As the student completes each assignment, this can be marked with an "X" in the box.

	Approximately five hours of course work a week
	Includes answer keys for worksheets, quizzes, lab reports, and exams
	Worksheets to help assess student learning
	Quizzes and exams are included to help reinforce learning and provide assessment opportunities
*	Designed for grades 9 to 12 in a one-year course

Course Objectives: Students completing this course will

- ✓ Investigate the core concepts of classical and modern biology
- Become familiar with the meaning of key terms in biology
- Explore the fundamental concepts of cell biology and important recent developments
- ✓ Study the development of classical genetics and modern concepts in the expression of DNA and how it varies with age and changing environmental conditions
- ✓ Learn the coordination and maintenance of the parts of an organism

- ✓ Study the variety of life forms and how their roles are integrated in an ecosystem
- ✓ Explore varied views of the origin of life and how the biblical narrative is consistent with what is observed in the study of biology
- Conduct laboratory procedures that develop important skills and illustrate important concepts
- Write lab reports each week to further develop writing skills in a scientific context

Course Description

This course provides important training and practice in developing skills involved in the study of biology, including observing and recognizing interactions and interdependencies of organisms in their natural environment, the use of a light microscope, dissection skills, and insights and recent advances in modern biology.

Life is from God. When God removes life, an organism dies. In Darwin's day, some organisms appeared to be very simple when contrasted to larger more complex organisms. Along with the blessings of modern tools to study life, it has become obvious that no life is simple. This makes sense because nothing that God creates is simple. An example is that the enzymes that control the functions of DNA are formed by the very DNA that they control. That would be like saying that a tool that you made helped you make the tool. How could a tool help you make itself if the tool were not there to begin with? Confusing, right? The DNA that controls all life forms was created at the beginning by the God that spoke life into being. The level of DNA necessary for life is called irreducible complexity.

The course begins with a review of chemical principles needed for biology, including the Biology of water, carbohydrates, proteins, and nucleic acids. The function of life is controlled by proteins called enzymes that are formed from the blueprint of DNA. Patterns of inheritance and modern genetics are introduced to help explain why we differ in so many ways. Patterns of variation in DNA and its expression are introduced. Consider that all of the variations in humans today had their roots in the DNA of two humans (Adam and Eve).

The course treats energy release through respiration, food and oxygen production through photosynthesis and chemosynthesis, along with the transfer of energy from producers to consumers to decomposers.

The kingdoms of life and their taxonomies are introduced along with experiences to study and observe various ecosystems and how they vary over time.

The course climaxes with consideration of issues of biological origins and how the approach differs depending upon assumptions of the nature of life and reality. Whether one comes from a theistic or secular point of view makes a huge difference in the interpretation of origins. The course concludes with human origins that have huge implications as to whether or not we were created in God's image with an eternal destiny or the sum product of natural laws acting upon atoms and molecules.

Each chapter in the textbook has accompanying worksheets and quizzes in the teacher's guide. Every 3 or 4 weeks, the student is to take an examination. Each chapter in the text has a laboratory exercise that teaches particular skills, illustrates insights in the concepts studied, and provides experience in preparing laboratory reports.

Teacher Instruction for this Course

This guide provides additional help in guiding students through the text and the laboratory exercises, along with a list of supplies not readily available around the house or in a local store. The labs are designed to be done at home. I have not included anything in the labs that I would not want in my home. The specimens for dissection are not preserved in harmful compounds — even though they should still be kept out of the mouth. Additional guidelines are given for the lab reports, worksheets, quizzes, and exams.

A mature study takes place in stages. Each of the concepts introduced in this study can be built upon and added to in future studies. How far a student should go in a study over a lifetime depends upon life goals. For example, this study may be sufficient for someone going into computer science, but someone going into medicine will take many more advanced courses later in college.

Reading: The required reading in this course is very rigorous and detailed. For students who are having difficulty in reading the chapter or understanding some of the concepts, the second day can be used to either complete the reading and/or worksheet or to focus on specific parts of the chapter that may need review. It is vital that the student understand the reading material to complete the course. The Glossary, which contains the vocabulary words, begins on page 320 in the student book.

Worksheets: The worksheets are important in indicating what the student needs to remember from the chapters and if they are ready for the quiz. The student needs to answer the questions in the worksheet. This is not a quiz, so the student can look at the answers after attempting them. The student needs to go back over the chapter to see why any questions were not answered correctly.

Quizzes: After understanding the correct answers to the questions on the worksheet, the student is to request the quiz. The quiz is to be taken with closed book and without using the worksheet. The teacher (parent) is to grade the quiz and record the grade on the grade sheet in the teacher's guide.

Exams: Every 3 or 4 weeks the student will take an exam. In preparation for the exam, the student is to reread the chapters covered on the exam, and answer the questions again on the worksheets and quizzes. This is why the answers were not written on the worksheets and quizzes. The student is to ask for the exam after studying the chapters, worksheets, and quizzes. The teacher (parent) grades the exam and records the grade.

It is normal to do better in some areas than in others. Learning is more than memorizing. Some of the questions depend upon recall, but some also depend upon reasoning skills. Through practice these skills are developed. If the student gets discouraged, be a source of encouragement. Some areas will be like going through a pleasant grassy valley, but others may be like fording a stream or climbing up a hillside.

The spiritual insights and Scripture references are important and not just a nice tack-on. These references are given in context with the subject matter. To just pop them in randomly would be to tell the student that they have nothing to do with biology — which is certainly not the case.

Vocabulary Words: On the second page of every chapter in the student book, vocabulary words are introduced that are highlighted in that chapter's text and have brief definitions found in the glossary at the back of the book. Students are encouraged to either write these out on 3 x 5 cards or to create another useful means of reviewing these throughout their course of study. Comprehension of sometimes difficult terms and concepts is very important to completing a course in Biology or any other complex science study.

Biology Credits for Transcripts

This is a one-year course with two full semesters, helping the student fulfill one credit per semester of biology, which includes the lab. High school transcripts will list the course as Biology with Labs. If questions arise from state agencies or schools, they can be referred to the course content.

Teacher Instructions for the Laboratory

This information is given so that the teacher (parent) can come alongside the student in helping line up the necessary materials, overseeing the procedures where necessary, and evaluating the lab reports. Be sure that complete sentences are used in the reports except where data are being recorded. This provides added writing experience and is clearer to someone reading the report. Do not hesitate to ask someone with more background in biology to come alongside in areas where you feel less confident.

Students are to use the "Laboratory" pages for taking notes, making general observations, and recording data. The "Laboratory Report" pages (see following examples) are for the student to write out their full observations, data, and conclusions. Any questions asked in the lab instructions are to be answered in the report. It may be helpful to read through the entire lab and the lab report before beginning each lab project.

WARNING: As with any science course that includes laboratory exercises, some things can be potentially hazardous if not handled properly. Make sure to carefully follow all instructions:

- ✓ Wear proper safety equipment when needed, including safety goggles/glasses.
- ▼ Keep small children away from where the labs are being conducted.
- ✓ Make sure clothing and other household items and surfaces are protected from staining.
- → Handle the microscope with care always carry it with two hands.
- ✓ Do not place anything used for labs in your mouth. Some chemicals can be poisonous.

Here is a list of supplies that need to be available for the labs. These are not readily available outside of a science laboratory and can be ordered online or purchased from a science supply retailer. Kits containing these supplies and a microscope can be obtained from Homeschool Science Products online.

Item	Order Info	Product ID	Lab Used
☐ 100 ml Beakers (3)	Supply Kit	CE-BEI0100	10, 26
☐ 1-2 foot square of screen / chicken wire	Hardware Stor	re	10, 26
☐ 2-inch long segments of water plant (5)	Pet Shop / Pond Algae	LD-ELODEA	12
☐ 50 ml graduated cylinder	Supply Kit	CE-CYGL050	2, 4, 7
☐ 95% Ethanol	Supply Kit	UN-ETHYALC	4
☐ Construction Paper Strips	Hobby Store		14
☐ Stopper (4)	Supply Kit	CE-STOP0XC	3
☐ Drinking Straw	Fast Food		12
☐ Earthworm Dissection Guide	Supply Kit	DE-GDWORM	24
☐ Earthworm Specimen	Supply Kit	PM-WORM	24
☐ Wax Pencil	Supply Kit	BE-WAXPENC	3, 7, 12, 13, 17
☐ Filter Funnel	Supply Kit	CE-FUNNEL	4
☐ Filter Paper	Supply Kit	CE-FILTPAP	4
☐ Frog Dissection Guide	Supply Kit	DE-GDFROG	25

Supply Kit	PM-FROG4S CH-DEXTROS CH-IODINE CH-METHBLU DE-SET1 MI-1100LED	25 13 3 6 3, 7, 13 5, 6, 8, 9, 14, 17,
Supply Kit Supply Kit Supply Kit Supply Kit	CH-IODINE CH-METHBLU DE-SET1	3 6 3, 7, 13
Supply Kit Supply Kit Supply Kit	CH-METHBLU DE-SET1	6 3, 7, 13
Supply Kit Supply Kit	DE-SET1	3, 7, 13
Supply Kit		
	MI-1100LED	5 6 8 9 14 17
Supply Kit		18, 22, 23, 24, 25
54PP7 1210	MS-SLIDSET	5, 6, 23
Supply Kit	GS-GLOVNIT	4, 24, 25
Supply Kit	CH-PHSTRIP	2
Supply Kit	CH-PHENRED	12
Supply Kit	DE-SET1	2, 3, 4, 6
Supply Kit	MS-SETBIO	6, 8, 9, 14, 17, 18, 22, 23, 24, 25
Supply Kit	CH-PTCTEST	19
Supply Kit	DE-SET1	24, 25
Supply Kit	DE-SET1	24, 25
Supply Kit	CE-STIR10	4
Supply Kit	DE-SET1	24, 25
Supply Kit	DE-TRAYS	24, 25
Supply Kit	CE-TTUBEKT	3, 4, 12, 13
Supply Kit	CE-TTUBESM	13
Supply Kit	DE-SET1	2, 18
lies		
Supply Kit	LM-ANTIKIT	17, 18
Supply Kit	LD-ESCHCOL	17, 18
Supply Kit	DE-GDGRASS	24
Supply Kit	PM-GRASHOP	24
Supply Kit	PT-SWABS	17
Supply Kit	BE-PPLATES	17, 18
	Supply Kit	Supply Kit CH-PHSTRIP Supply Kit DE-SET1 Supply Kit DE-TRAYS Supply Kit DE-TRAYS Supply Kit CE-TTUBEKT Supply Kit CE-TTUBESM Supply Kit DE-SET1 Supply Kit DE-SET1 Supply Kit DE-SET1 LM-ANTIKIT Supply Kit LD-ESCHCOL Supply Kit DE-GDGRASS Supply Kit PM-GRASHOP Supply Kit PT-SWABS

Note: *Order these items 7–10 days before you need them as they are perishable.

Laboratory Report (20 points possible)

Living Things Observed in the Area

At the Bozeman ponds (Bozeman, Montana) I see a body of water surrounded by tall willow trees and an occasional path leading down to the water. There are grasses farther on shore from the trees. The water is rippling with mallard ducks swimming across the water and on the shore. There is a lot of leaf litter on the shore and in the water from the trees.

A Keystone Species

The willow trees appear to be a keystone species. They stabilize the shoreline and provide shelter for the ducks. The leaves appear to be discolored perhaps with fungus and bacteria growing on them.

I cannot see any water insects or algae in the water, but the ducks are feeding on something in the water and these are typically fed upon by ducks. The ducks will deposit their waste in the water and on the shore which will provide nutrients for the algae, water insects, and shore plants. The ducks also appear to be a keystone species.

Life Forms That Feed Upon Other Life Forms

The ducks appear to feed on water insects and algae in the water.

Life Forms That Are Eaten by Other Life Forms

Water insects and algae appear to be eaten by the ducks.

What decomposers are in the area?

Bacteria and algae appear to be decomposing the leaf litter on the shore and in the water. This could be shown by the discoloration of the leaves which are slowly coming apart.

Sample Biology Lab Reports Some of the numbers have been changed from what is asked for in the lab so that the students have to do their own work.

pH and Hydrogen Bonding of Water Molecules

Day 5

Lesson 2 Sample Laboratory Report Name

Laboratory Report (20 points possible)

Procedure and Results

The pH of a solution

- 1. The pH of the distilled water is 7.
- 2. After adding 5 drops of vinegar to the distilled water, the pH is 6.
- 3. After adding 5 more drops of vinegar to the water from step 2, the pH is 4.
- 4. After adding an additional 5 drops of vinegar to the water from step 3, the pH is 2.
- 5. As more acid (vinegar) was added to the distilled water, the pH dropped, indicating increased acidity.
- 6. As 1/2 teaspoon of sodium bicarbonate was added to the acidic solution from step 4, the water foamed. The pH of the resulting solution was 8, indicating that the sodium bicarbonate decreased the acidity and increased the basic nature of the solution as indicated by the pH increase from 2 to 8.

Hydrogen bonds between water molecules

- 1. Water was added to a drinking glass carefully using a pipette until there was a slight mound in the middle of the water (meniscus).
- 2. I carefully laid a sewing needle on top of the water in the middle of the meniscus using a pair of tweezers. The sewing needle floated on top of the water because the hydrogen bonds between the water molecules caused the water molecules to "stick together" under the needle.
- 3. I added a drop of liquid dish soap to the water and the needle sank. It appeared that the soap disrupted the hydrogen bonds between the water molecules and the needle sank. The top layer of the water flattened out and the meniscus was disrupted.

Sample Biology Lab Reports Some of the numbers have been changed from what is asked for in the lab so that the students have to do their own work.

Teacher Instructions for Quizzes and Examinations

Teacher's Instructions for Quizzes

The quizzes are to be given at the end of the study as per the schedule of the lessons. The students are to review the text of the lesson and the worksheets. Grade the quiz from the answers in the teacher's guide. Have the student look up any questions that were missed and explain to you what the correct answer should be and why. The quizzes are multiple choice and matching (with few exceptions) to make grading easier on your part. There are 28 quizzes with 15 points possible for each quiz. This gives a possible total of 420 points. The customary grading scale is:

90%–100% is an A;
80%–89% is a B;
70%–79% is a C;
60%–69% is a D and
59% and lower is an F.

In science studies, an A and B are very good. C is average.
D or F indicates the need for more maturity, more practice, or more study. Future success is always possible with maturity, study, and practice.

This applies to each individual quiz. At the end of the course, the average of the quizzes is to be added to the average of the exams to give a final score graded according to this scale. If a student misses more than 50% on a quiz, the quiz is to be retaken after careful study. You can give the student back ½ point for each answer gotten correct the second time that was missed the first time. This can be done for up to 5 quizzes. I have found this policy to be very helpful for students that get off to a slow start. As well, I have found that a student's readiness for a study of this nature depends more upon maturity rather than age. Always encourage your student but still hold the standard and do not cut corners. That way the student will have the assurance of being able to go on to further studies and succeed.

There is no midterm or final examination because by its very nature, Biology is comprehensive. The concepts learned earlier are used in the later lessons and labs throughout the course.

Teacher's Instructions for Administering the Examinations

In the week of an examination, the student is to study the previous quizzes and the practice exercises for the lessons covered on the exam. The exam is like an expanded version of a quiz. Each exam consists of 30 multiple choice or matching questions (with few exceptions). The questions are not verbatim from the quizzes but are similar. An examination is a sampling of the material and does not include every point covered in the lessons.

A high school transcript usually has 1 grade for science courses (lab and lecture combined) and so this would appear as 1 credit with labs in Biology. (Note that some states may calculate credits in a different manner.) This can be determined by making the quizzes and exams 75 percent of the grade and the lab 25 percent of the grade. To find the lab grade take the total points earned from all of the labs divided by the total possible times 100. An example of finding the total grade is if the average of the quizzes and exams are 85 percent and the labs are 97 percent:

Quiz/Exam Average <u>85</u> x 3 = <u>255</u> + Lab <u>97</u> = <u>352</u> / 4 x 100 = <u>88% (B+)</u> **Final Grade**

Grading Sheet

Lesson	Quiz	Exam		Lab
Lesson 1	/ 15			/ 20
Lesson 2	/ 15			/ 20
Lesson 3	/ 15			/ 20
Lesson 4	/ 15	Examination 1	/ 30	/ 20
Lesson 5	/ 15			/ 20
Lesson 6	/ 15			/ 20
Lesson 7	/ 15			/ 20
Lesson 8	/ 15			/ 20
Lesson 9	/ 15	Examination 2	/ 30	/ 20
Lesson 10	/ 15			/ 20
Lesson 11	/ 15			/ 20
Lesson 12	/ 15	Examination 3	/ 30	/ 20
Lesson 13	/ 15			/ 20
Lesson 14	/ 15			/ 20
Lesson 15	/ 15			/ 20
Lesson 16	/ 15	Examination 4	/ 30	/ 20
Lesson 17	/ 15			/ 20
Lesson 18	/ 15			/ 20
Lesson 19	/ 15	Examination 5	/ 30	/ 20
Lesson 20	/ 15			/ 20
Lesson 21	/ 15			/ 20
Lesson 22	/ 15	Examination 6	/ 30	/ 20
Lesson 23	/ 15			/ 20
Lesson 24	/ 15			/ 20
Lesson 25	/ 15	Examination 7	/ 30	/ 20
Lesson 26	/ 15			/ 20
Lesson 27	/ 15			/ 20
Lesson 28	/ 15	Examination 8	/ 30	/ 20
Total Score / Percent	/ 420 = %	/ 240 =	%	/ 560 = %
	Quizzes % + Exan	ninations % / 2 =	=%	%
Quiz/Exan	n Average x 3 =	+ Lab =	/ 4 x 100 =	Final Grade

First Semester Suggested Daily Schedule

Date	Day	Assignment	Due Date	√	Grade
		First Semester-First Quarter			
	Day 1	Begin Chapter 1 Chemical Principles in Biology • Read Pages 4–15 • Biology • (BIO) Begin Lesson 1 Worksheet 1 Pages 25–26 • Teacher Guide • (TG)			
Week 1	Day 2	Complete the reading and worksheet for Chapter 1; Focus on Vocabulary Words Page 6 • (BIO)			
	Day 3	Review Chapter 1 • Complete Quiz 1 Pages 239-240 • (TG)			
	Day 4	Read Laboratory 1 Biology — The Study of Life Pages 16–17 • (BIO) • Start Laboratory 1 Page 27 • (TG)			
	Day 5	Conclude Laboratory 1 and Prepare Lesson 1 Lab Report Page 29 • (TG)			
	Day 6	Begin Chapter 2 Water • Read Pages 18–23 • (BIO) Begin Lesson 2 Worksheet 1 Page 31 • (TG)			
	Day 7	Complete the reading and worksheet for Chapter 2; Focus on Vocabulary Words Page 20 • (BIO)			
Week 2	Day 8	Review Chapter 2 • Complete Quiz 2 Pages 241-242 • (TG)			
	Day 9	Read Laboratory 2 pH and Hydrogen Bonding of Water Molecules Pages 24–25 • (BIO) • Start Laboratory 2 Pages 33-34 • (TG)			
	Day 10	Conclude Laboratory 2 and Prepare Lesson 2 Lab Report Page 35 • (TG)			
	Day 11	Begin Chapter 3 Carbohydrates and Lipids • Read Pages 26–31 • (BIO) • Begin Lesson 3 Worksheet 1 Pages 37-38 • (TG)			
	Day 12	Complete the reading and worksheet for Chapter 3; Focus on Vocabulary Words Page 28 • (BIO)			
Week 3	Day 13	Review Chapter 3 • Complete Quiz 3 Pages 243-244 • (TG)			
	Day 14	Read Laboratory 3 Carbohydrates Pages 32–33 • (BIO) Start Laboratory 3 Pages 39-40 • (TG)			
	Day 15	Conclude Laboratory 3 and Prepare Lesson 3 Lab Report Pages 41-42 • (TG)			
	Day 16	Begin Chapter 4 Proteins and Nucleic Acids • Read Pages 34-41 • (BIO) • Begin Lesson 4 Worksheet 1 Pages 43-44 • (TG)			
	Day 17	Complete the reading and worksheet for Chapter 4; Focus on Vocabulary Words Page 36 • (BIO)			
Week 4	Day 18	Review Chapter 4 • Complete Quiz 4 Pages 245-246 • (TG)			
	Day 19	Read Laboratory 4 Nucleic Acids Pages 42–43 • (BIO) Start Laboratory 4 Pages 45-46 • (TG)			
	Day 20	Conclude Laboratory 4 and Prepare Lesson 4 Lab Report Page 47 • (TG)			
	Day 21	Review Lesson 1 and Lesson 1 Quiz			
	Day 22	Review Lesson 2 and Lesson 2 Quiz			
Week 5	Day 23	Review Lesson 3 and Lesson 3 Quiz			
	Day 24	Review Lesson 4 and Lesson 4 Quiz			
	Day 25	Take Exam 1 (Lessons 1–4) Pages 297–298 • (TG)			

Date	Day	Assignment	Due Date	\checkmark	Grade
	Day 26	Begin Chapter 5 Nature of Cells • Read Pages 44–51 • (BIO) Begin Lesson 5 Worksheet 1 Pages 49-50 • (TG)			
	Day 27	Complete the reading and worksheet for Chapter 5; Focus on Vocabulary Words Page 46 • (BIO)			
Week 6	Day 28	Review Chapter 5 • Complete Quiz 5 Page 247 • (TG)			
	Day 29	Read Laboratory 5 Use of the Microscope Pages 52-55 • (BIO) Start Laboratory 5 Pages 51-53 • (TG)			
	Day 30	Conclude Laboratory 5 and Prepare Lesson 5 Lab Report Page 55 • (TG)			
	Day 31	Begin Chapter 6 Cell Membranes and Nucleus • Read Pages 56–61 • (BIO) Begin Lesson 6 Worksheet 1 Page 57 • (TG)			
**** 1 =	Day 32	Complete the reading and worksheet for Chapter 6; Focus on Vocabulary Words Page 58 • (BIO)			
Week 7	Day 33	Review Chapter 6 • Complete Quiz 6 Pages 249-250 • (TG)			
	Day 34	Read Laboratory 6 Cells Pages 62-65 • (BIO) Start Laboratory 6 Pages 59-61 • (TG)			
	Day 35	Conclude Laboratory 6 and Prepare Lesson 6 Lab Report Page 63 • (TG)			
	Day 36	Begin Chapter 7 Movement Through Cell Membranes • Read Pages 66–71 • (BIO) Begin Lesson 7 Worksheet 1 Page 65 • (TG)			
**** 1 0	Day 37	Complete the reading and worksheet for Chapter 7; Focus on Vocabulary Words Page 68 • (BIO)			
Week 8	Day 38	Review Chapter 7 • Complete Quiz 7 Pages 251-252 • (TG)			
	Day 39	Read Laboratory 7 Osmosis Pages 72–73 • (BIO) Start Laboratory 7 Pages 67-68 • (TG)			
	Day 40	Conclude Laboratory 7 and Prepare Lesson 7 Lab Report Pages 69-70 • (TG)			
	Day 41	Begin Chapter 8 Cell Organelles • Read Pages 74-81 • (BIO) Begin Lesson 8 Worksheet 1 Pages 71-72 • (TG)			
	Day 42	Complete the reading and worksheet for Chapter 8; Focus on Vocabulary Words Page 76 • (BIO)			
Week 9	Day 43	Review Chapter 8 • Complete Quiz 8 Pages 253-254 • (TG)			
	Day 44	Read Laboratory 8 Cell Structures Pages 82-85 • (BIO) Start Laboratory 8 Pages 73-74 • (TG)			
	Day 45	Conclude Laboratory 8 and Prepare Lesson 8 Lab Report Pages 75-76 • (TG)			

Date	Day	Assignment	Due Date	\checkmark	Grade
		First Semester-Second Quarter			
	Day 46	Review Lesson 5 and Lesson 5 Quiz			
	Day 47	Review Lesson 6 and Lesson 6 Quiz			
Week 1	Day 48	Review Lesson 7 and Lesson 7 Quiz			
	Day 49	Review Lesson 8 and Lesson 8 Quiz			
	Day 50	Take Exam 2 (Lessons 5–8) Pages 299-300 • (TG)			
	Day 51	Begin Chapter 9 Cell Division • Read Pages 86-91 • (BIO) Begin Lesson 9 Worksheet 1 Pages 77-78 • (TG)			
	Day 52	Complete the reading and worksheet for Chapter 9; Focus on Vocabulary Words Page 88 • (BIO)			
Week 2	Day 53	Review Chapter 9 • Complete Quiz 9 Page 255 • (TG)			
	Day 54	Read Laboratory 9 Cell Division Pages 92–93 • (BIO) Start Laboratory 9 Pages 79-80 • (TG)			
	Day 55	Conclude Laboratory 9 and Prepare Lesson 9 Lab Report Page 81 • (TG)			
	Day 56	Begin Chapter 10 Ecosystems • Read Pages 94–101 • (BIO) Begin Lesson 10 Worksheet 1 Page 83 • (TG)			
	Day 57	Complete the reading and worksheet for Chapter 10; Focus on Vocabulary Words Page 96 • (BIO)			
Week 3	Day 58	Review Chapter 10 • Complete Quiz 10 Page 257 • (TG)			
	Day 59	Read Laboratory 10 Ecosystems Pages 102–103 • (BIO) Start Laboratory 10 Pages 85-86 • (TG)			
	Day 60	Conclude Laboratory 10 and Prepare Lesson 10 Lab Report Pages 87-88 • (TG)			
	Day 61	Begin Chapter 11 Biomes • Read Pages 104-117 • (BIO) Begin Lesson 11 Worksheet 1 Pages 89-90 • (TG)			
	Day 62	Complete the reading and worksheet for Chapter 11; Focus on Vocabulary Words Page 106 • (BIO)			
Week 4	Day 63	Review Chapter 11 • Complete Quiz 11 Page 259 • (TG)			
	Day 64	Read Laboratory 11 Biomes Pages 118-119 • (BIO) Start Laboratory 4 Page 91 • (TG)			
	Day 65	Conclude Laboratory 11 and Prepare Lesson 11 Lab Report Page 93 • (TG)			
	Day 66	Begin Chapter 12 Energy Capture — Photosynthesis • Read Pages 120-127 • (BIO) Begin Lesson 12 Worksheet 1 Pages 95–96 • (TG)			
	Day 67	Complete the reading and worksheet for Chapter 12; Focus on Vocabulary Words Page 122 • (BIO)			
Week 5	Day 68	Review Chapter 12 • Complete Quiz 12 Page 261 • (TG)			
	Day 69	Read Laboratory 12 Photosynthesis Pages 128-129 • (BIO) Start Laboratory 12 Pages 97-98 • (TG)			
	Day 70	Conclude Laboratory 12 and Prepare Lesson 12 Lab Report Page 99 • (TG)			

Date	Day	Assignment	Due Date	\checkmark	Grade
	Day 71	Review Lesson 9 and Lesson 9 Quiz			
Week 6	Day 72	Review Lesson 10 and Lesson 10 Quiz			
	Day 73	Review Lesson 11 and Lesson 11 Quiz			
	Day 74	Review Lesson 12 and Lesson 12 Quiz			
	Day 75	Take Exam 3 (Lessons 9–12) Pages 301-302 • (TG)			
	Day 76	Begin Chapter 13 Energy Release — Respiration • Read Pages 130-137 • (BIO) Begin Lesson 13 Worksheet 1 Pages 101-103 • (TG)			
W 1 =	Day 77	Complete the reading and worksheet for Chapter 13; Focus on Vocabulary Words Page 132 • (BIO)			
Week 7	Day 78	Review Chapter 13 • Complete Quiz 13 Page 263 • (TG)			
	Day 79	Read Laboratory 13 Cellular Respiration Pages 138-139 • (BIO) Start Laboratory 13 Pages 105-106 • (TG)			
	Day 80	Conclude Laboratory 13 and Prepare Lesson 13 Lab Report Page 107 • (TG)			
	Day 81	Begin Chapter 14 Chromosomes and Genes • Read Pages 140-145 • (BIO) • Begin Lesson 14 Worksheet 1 Pages 109-110 • (TG)			
	Day 82	Complete the reading and worksheet for Chapter 14; Focus on Vocabulary Words Page 142 • (BIO)			
Week 8	Day 83	Review Chapter 14 • Complete Quiz 14 Pages 265-266 • (TG)			
	Day 84	Read Laboratory 14 Chromosomes and Genes Pages 146-149 • (BIO) • Start Laboratory 14 Pages 111-113 • (TG)			
	Day 85	Conclude Laboratory 14 and Prepare Lesson 14 Lab Report Page 115 • (TG)			
	Day 86	Begin Chapter 15 The Genetic Code • Read Pages 150-155 • (BIO) Begin Lesson 15 Worksheet 1 Pages 117-118 • (TG)			
Week 9	Day 87	Complete the reading and worksheet for Chapter 15; Focus on Vocabulary Words Page 152 • (BIO)			
	Day 88	Review Chapter 15 • Complete Quiz 15 Page 267 • (TG)			
	Day 89	Read Laboratory 15 The Genetic Code Pages 156-157 • (BIO) Start Laboratory 15 Pages 119-120 • (TG)			
	Day 90	Conclude Laboratory 15 and Prepare Lesson 15 Lab Report Page 121 • (TG)			
		Mid-Term Grade			

Second Semester Suggested Daily Schedule

Date	Day	Assignment	Due Date	√	Grade
		Second Semester-Third Quarter			
Week 1	Day 91	Chapter 16 Expression of DNA — Transcription • Read Pages 158-163 • (BIO) Begin Lesson 16 Worksheet 1 Pages 123-124 • (TG)			
	Day 92	Complete the reading and worksheet for Chapter 16; Focus on Vocabulary Words Page 160 • (BIO)			
	Day 93	Review Chapter 16 • Complete Quiz 16 Page 269 • (TG)			
	Day 94	Read Laboratory 16 Transcription — mRNA Pages 164-167 • (BIO) • Start Laboratory 16 Pages 125-128 • (TG)			
	Day 95	Conclude Laboratory 16 and Prepare Lesson 16 Lab Report Page 129 • (TG)			
	Day 96	Review Lesson 13 and Lesson 13 Quiz			
	Day 97	Review Lesson 14 and Lesson 14 Quiz			
Week 2	Day 98	Review Lesson 15 and Lesson 15 Quiz			
	Day 99	Review Lesson 16 and Lesson 16 Quiz			
	Day 100	Take Exam 4 (Lessons 13–16) Pages 303-304 • (TG)			
	Day 101	Chapter 17 Expression of DNA — Translation • Read Pages 168-173 • (BIO) Begin Lesson 17 Worksheet 1 Pages 131-132 • (TG)			
	Day 102	Complete the reading and worksheet for Chapter 17; Focus on Vocabulary Words Page 170 • (BIO)			
Week 3	Day 103	Review Chapter 17 • Complete Quiz 17 Page 271 • (TG)			
	Day 104	Read Laboratory 17 Disruption of DNA Translation and Products of DNA Translation within Bacteria Cells Pages 174-177 • (BIO) • Start Laboratory 17 Pages 133-136 • (TG)			
	Day 105	Conclude Laboratory 17 and Prepare Lesson 17 Lab Report Page 137 • (TG)			
	Day 106	Chapter 18 Perpetuation of Life • Read Pages 178-185 • (BIO) Begin Lesson 18 Worksheet 1 Pages 139-140 • (TG)			
	Day 107	Complete the reading and worksheet for Chapter 18; Focus on Vocabulary Words Page 180 • (BIO)			
Week 4	Day 108	Review Chapter 18 • Complete Quiz 18 Page 273 • (TG)			
week 4	Day 109	Read Laboratory 18 Diverse Products of Protein Translation at Different Stages of Development Pages 186-189 • (BIO) Start Laboratory 18 Pages 141-143 • (TG)			
	Day 110	Conclude Laboratory 18 and Prepare Lesson 18 Lab Report Page 145 • (TG)			

Date	Day	Assignment	Due Date	\checkmark	Grade
	Day 111	Chapter 19 Genetics Patterns I • Read Pages 190-195 • (BIO) Begin Lesson 19 Worksheet 1 Pages 147-148 • (TG)			
	Day 112	Complete the reading and worksheet for Chapter 19; Focus on Vocabulary Words Page 192 • (BIO)			
Week 5	Day 113	Review Chapter 19 • Complete Quiz 19 Page 275 • (TG)			
	Day 114	Read Laboratory 19 Human Genetics Pages 196-199 • (BIO) Start Laboratory 19 Pages 149-152 • (TG)			
	Day 115	Conclude Laboratory 19 and Prepare Lesson 19 Lab Report Pages 153-155 • (TG)			
	Day 116	Review Lesson 17 and Lesson 17 Quiz			
	Day 117	Review Lesson 18 and Lesson 18 Quiz			
Week 6	Day 118	Review Lesson 19 and Lesson 19 Quiz			
	Day 119	Review Lessons 17–19			
	Day 120	Take Exam 5 (Lessons 17–19) Pages 305-306 • (TG)			
	Day 121	Chapter 20 Genetics Patterns 2 • Read Pages 200-207 • (BIO) Begin Lesson 20 Worksheet 1 Pages 157-158 • (TG)			
	Day 122	Complete the reading and worksheet for Chapter 20; Focus on Vocabulary Words Page 202 • (BIO)			
Week 7	Day 123	Review Chapter 20 • Complete Quiz 20 Page 277 • (TG)			
	Day 124	Read Laboratory 20 Dihybrid Test Cross with Corn Pages 208-211 • (BIO) • Start Laboratory 20 Pages 159-162 • (TG)			
	Day 125	Conclude Laboratory 20 and Prepare Lesson 20 Lab Report Page 163 • (TG)			
	Day 126	Chapter 21 Genetic Mutations and Variations • Read Pages 212-219 • (BIO) Begin Lesson 21 Worksheet 1 Pages 165-166 • (TG)			
	Day 127	Complete the reading and worksheet for Chapter 21; Focus on Vocabulary Words Page 214 • (BIO)			
Week 8	Day 128	Review Chapter 21 • Complete Quiz 21 Page 279 • (TG)			
	Day 129	Read Laboratory 21 Genetic Variation Pages 220-223 • (BIO) Start Laboratory 21 Pages 167-169 • (TG)			
	Day 130	Conclude Laboratory 21 and Prepare Lesson 21 Lab Report Page 171 • (TG)			
	Day 131	Chapter 22 Genomics • Read Pages 224-229 • (BIO) Begin Lesson 22 Worksheet 1 Page 173 • (TG)			
	Day 132	Complete the reading and worksheet for Chapter 22; Focus on Vocabulary Words Page 226 • (BIO)			
Week 9	Day 133	Review Chapter 22 • Complete Quiz 22 Page 281 • (TG)			
	Day 134	Read Laboratory 22 Plants Pages 230-233 • (BIO) Start Laboratory 22 Pages 175-177 • (TG)			
	Day 135	Conclude Laboratory 22 and Prepare Lesson 22 Lab Report Page 179 • (TG)			

Date	Day	Assignment	Due Date	\checkmark	Grade
		Second Semester-Fourth Quarter			
	Day 136	Review Lesson 20 and Lesson 20 Quiz			
	Day 137	Review Lesson 21 and Lesson 21 Quiz			
Week 1	Day 138	Review Lesson 22 and Lesson 22 Quiz			
	Day 139	Review Lessons 20–22			
	Day 140	Take Exam 6 (Lessons 20–22) Pages 307-308 • (TG)			
	Day 141	Chapter 23 Plant Taxonomy • Read Pages 234-243 • (BIO) Begin Lesson 23 Worksheet 1 Pages 181-182 • (TG)			
	Day 142	Complete the reading and worksheet for Chapter 23; Focus on Vocabulary Words Page 236 • (BIO)			
Week 2	Day 143	Review Chapter 23 • Complete Quiz 23 Page 283 • (TG)			
	Day 144	Read Laboratory 23 Protistans and Fungi Pages 244-247 • (BIO) Start Laboratory 23 Pages 183-186 • (TG)			
	Day 145	Conclude Laboratory 23 and Prepare Lesson 23 Lab Report Page 187 • (TG)			
	Day 146	Chapter 24 Animal Taxonomy — Invertebrates • Read Pages 248-257 • (BIO) Begin Lesson 24 Worksheet 1 Pages 189-191 • (TG)			
**** 1 -	Day 147	Complete the reading and worksheet for Chapter 24; Focus on Vocabulary Words Page 250 • (BIO)			
Week 3	Day 148	Review Chapter 24 • Complete Quiz 24 Page 285 • (TG)			
	Day 149	Read Laboratory 24 Invertebrate Animals Pages 258-263 • (BIO) Start Laboratory 24 Pages 193-198 • (TG)			
	Day 150	Conclude Laboratory 24 and Prepare Lesson 24 Lab Report Page 199 • (TG)			
	Day 151	Chapter 25 Animal Taxonomy — Vertebrates • Read Pages 264-277 • (BIO) Begin Lesson 25 Worksheet 1 Pages 201-203 • (TG)			
**** 1 /	Day 152	Complete the reading and worksheet for Chapter 25; Focus on Vocabulary Words Page 266 • (BIO)			
Week 4	Day 153	Review Chapter 25 • Complete Quiz 25 Page 287 • (TG)			
	Day 154	Read Laboratory 25 Vertebrate Animals Pages 278-283 • (BIO) Start Laboratory 25 Pages 205-208 • (TG)			
	Day 155	Conclude Laboratory 25 and Prepare Lesson 25 Lab Report Page 209 • (TG)			
	Day 156	Review Lesson 23 and Lesson 23 Quiz			
	Day 157	Review Lesson 24 and Lesson 24 Quiz			
Week 5	Day 158	Review Lesson 25 and Lesson 25 Quiz			
	Day 159	Review Lessons 23–25			
	Day 160	Take Exam 7 (Lessons 23–25) Pages 309-310 • (TG)			

Date	Day	Assignment	Due Date	\checkmark	Grade
	Day 161	Chapter 26 Views of Biological Origins • Read Pages 284-295 • (BIO) • Begin Lesson 26 Worksheet 1 Pages 211-213 • (TG)			
	Day 162	Complete the reading and worksheet for Chapter 26; Focus on Vocabulary Words Page 286 • (BIO)			
Week 6	Day 163	Review Chapter 26 • Complete Quiz 26 Page 289 • (TG)			
	Day 164	Read Laboratory 26 Sedimentation and Fossils Pages 296-297 • (BIO) • Start Laboratory 26 Pages 215-217 • (TG)			
	Day 165	Conclude Laboratory 26 and Prepare Lesson 26 Lab Report Page 219 • (TG)			
	Day 166	Chapter 27 Evidences of Biological Origins • Read Pages 298-305 • (BIO) Begin Lesson 27 Worksheet 1 Pages 221-223 • (TG)			
	Day 167	Complete the reading and worksheet for Chapter 27; Focus on Vocabulary Words Page 300 • (BIO)			
Week 7	Day 168	Review Chapter 27 • Complete Quiz 27 Page 291 • (TG)			
	Day 169	Read Laboratory 27 Diversity Within Kinds of Creation Pages 306-307 • (BIO) • Start Laboratory 27 Pages 225-226 • (TG)			
	Day 170	Conclude Laboratory 27 and Prepare Lesson 27 Lab Report Page 227 • (TG)			
	Day 171	Chapter 28 Human Origins • Read Pages 308-315 • (BIO) Begin Lesson 28 Worksheet 1 Pages 229-231 • (TG)			
	Day 172	Complete the reading and worksheet for Chapter 28; Focus on Vocabulary Words Page 310 • (BIO)			
Week 8	Day 173	Review Chapter 28 • Complete Quiz 28 Page 293 • (TG)			
	Day 174	Read Laboratory 28 Human Origins Pages 316-319 • (BIO) Start Laboratory 28 Pages 233-234 • (TG)			
	Day 175	Conclude Laboratory 28 and Prepare Lesson 28 Lab Report Page 235 • (TG)			
	Day 176	Review Lesson 26 and Lesson 26 Quiz			
	Day 177	Review Lesson 27 and Lesson 27 Quiz			
Week 9	Day 178	Review Lesson 28 and Lesson 28 Quiz			
	Day 179	Review Lessons 26–28			
	Day 180	Take Exam 8 (Lessons 26–28) Pages 311-312 • (TG)			
		Final Grade			

Worksheets and Laboratory Reports



Chemical Principles in Biology

Day 1

Lesson 1 Worksheet 1 Name

Fill in the Blank

1. Al	ll atoms with the same chemical behavior are the same
	he major building blocks of atoms are,, and
	he atomic number of an atom is its number of
4. Aı	toms with the same atomic number are the same
5. Aı	toms bonded together form
6. W	Trite out the chemical equation for photosynthesis.
7 . _	molecules of CO_2 are used to make one molecule of glucose.
8. Fo	or every molecule of glucose formed in photosynthesis, $\underline{\hspace{1cm}}$ molecules of O_2 are formed.
9. D	raw the structural formula of glucose.
10.	
11. Tł	here are H atoms in one glucose molecule.
12. Tł	here are C atoms in one glucose molecule.
13	and are found in the nucleus of an atom.
14. A	proton has a charge of
15. A	neutron has a charge of
16. Aı	n electron has a charge of
17. Tł	he mass of an atom is determined by its and
	n atoms atomic mass number is the sum of the number of its and
19. Iso	otopes are atoms that have the same number of but different numbers of
	N ₇ has protons and neutrons.
21. Aı	re ${}^{14}C_6$ and ${}^{12}C_6$ isotopes of each other? Why?

23. Erwin Schrodinger in 1926 described the electrons of atoms by their energies rather than their physica positions. He called these energy levels
24. According to his model, there could be electrons in the lowest energy level.
25. An atom would be more stable if there were a total of electrons in the next energy level. An atom would gain or lose electrons to achieve this number of electrons.
26. Metal atoms like sodium and calcium tend to electrons to achieve this number.
27. Non-metal atoms like oxygen and chlorine tend to electrons to achieve this number.
28. When a sodium (metal) atom forms a bond with a chlorine (non-metal) atom, they formbonds.
29. This is an rather than a true physical bond.
30. The charged sodium ion attracts the charged chloride
ion.
31. When two non-metal atoms (such as two oxygen atoms) bond together, they
electrons forming bonds.

32. Carbon atoms with six electrons have ______ electrons in the lowest energy level and ______ electrons in the next higher energy level. They each need _____ electrons to make _____

22. Are ¹⁴C₆ and ¹⁴N₇ isotopes of each other? Why?

electrons in the next higher energy level.

33. How many chemical bonds are there for each carbon atom in the glucose molecule?

Day 4

Lesson 1 Laboratory Name

Laboratory 1: Biology — The Study of Life	Lab Notes:
REQUIRED MATERIALS	
☐ Pencil and paper	
INTRODUCTION	
This exercise sets the stage for the rest of this study. Biology is the study of the living. Think back to this laboratory exercise to keep a right perspective while doing the other laboratory exercises. The abundant forms and quantity of life that God has created and their relationships to each other is the focus of this study.	
PURPOSE	
To gain an overview perspective for the study of biology.	
PROCEDURE	
Go outside; find a quiet place to observe the life around you. Take a few minutes to just look at everything around you.	
Make a list and brief description of most of the living things that you see. Use complete sentences so that it is meaningful to whoever reads this report. Write in such a way that someone else can picture in their mind what you are describing.	
Identify what you could consider to be a keystone species. This is a life form that if removed would have an effect on the other life forms around it. An example would be a pine tree where squirrels get seeds for food from the pinecones. Describe why you consider it a keystone species.	
Describe a life form that you see that feeds on another life form.	
Describe a life form that you see that is food for another life form.	
Dead organisms and waste products have to be broken down so that they can become nutrients for others (such as plants). Describe which life forms you see or know to be there (some are there even though you cannot see them) that carry out this role. These are called decomposers — usually bacteria and fungi.	
Notes may be taken on the right of each lab, but complete sentences and all information need to be filled out on the actual Laboratory Report sheet.	



Biology

Biology — The Study of Life

Day 5

Lesson 1
Laboratory Report

Name

Laboratory Report (20 points possible)

1. Living things observed in the area

2. Keystone species

3. Life forms that feed upon other life forms

4. Life forms that are fed upon by other life forms

5. Decomposers in the area



Water

Day 6

Lesson 2 Worksheet 1 Name

Fill in the Blank

1.	A water molecule has a partial charge on the oxygen side and a partial
	charge on the side of the hydrogen atoms.
2.	This causes the hydrogen side of the water molecule to be to the oxygen side of another water molecule.
3.	This weak attraction is called a bond.
4.	The oxygen ends of water molecules are charged and the hydrogen ends of water molecules are charged.
5.	Water molecules are polar and oil is
6.	The word for water loving is and the word for water fearing is
7.	A calorie is the amount of heat energy necessary to raise of water one degree Celsius.
8.	Water has a heat capacity, meaning that it takes heat to boil water.
9.	Ice is dense than liquid water which is why ice in liquid water.
10.	If it were not for the bonds between water molecules, water would boil at less than °F.
11.	Water dissolves polar compounds because of its bonds.
12.	About 1 out of every water molecules comes apart as and
13.	is an acid.
14.	is a base.
15.	pH = 7 is
16.	pH = 9 has x more than pH = 7.
17.	pH = 6 has x more than pH = 7.

Day 9

Lesson 2 Laboratory Name

Lab 2 — pH and Hydrogen Bonding of Water Molecules	Lab Notes:		
REQUIRED MATERIALS			
☐ Distilled water (local from grocery store)☐ Vinegar			
□ Baking powder□ Drinking glass with a smooth rim			
☐ Sewing needle☐ Liquid soap			
☐ Three 100 ml beakers (from the supply kit)			
 50 ml graduated cylinder (from the supply kit) pH indicator paper (from the supply kit) Eyedropper (from the supply kit) 			
INTRODUCTION			
One out of every 10 million (10^7) water molecules comes apart to give an acid (H^+) and a base (^-OH).			
$H_2O \leftrightarrows H^+ + {}^-OH$			
A water molecule yields 1 H ⁺ and 1 ⁻ OH so the amount of acid and base equals each other. Water is said to have neutral pH with a pH of 7. The 7 comes from the expression 10 ⁷ . If more acid is added, the solution is acidic and the pH is below 7. If more base is added, the solution is basic and the pH is greater than 7.			
Water molecules have a partial positive charge on the hydrogen side of the molecule and a partial negative charge on the			
oxygen side of the molecule. This causes the positive sides of water molecules to attract the negative sides of other molecules			
producing the hydrogen bonds between water molecules. Even			
though these bonds are very weak, the attraction is appreciable because there are so many of them. Water molecules are so light			
in weight that if it were not for the hydrogen bonds between them, they would evaporate and boil at less than -200° F. If			
that were the case, biological life would be impossible on earth. What an example of God's grace!			
PURPOSE This oversies involves the measurement of a solution's pH and			
This exercise involves the measurement of a solution's pH and how it is changed. The second part of this exercise demonstrates			
the effect of the hydrogen bonds between water molecules.			

PR	OCEDURE	
	swer the questions in this lab in your lab report with applete sentences.	
	e pH of a solution is measured by the reaction of H ⁺ ions d ⁻ OH ions with indicator compounds in pH paper.	
1.	Add 50 ml (milliliters measured with a 50 ml graduated cylinder) of distilled water to a 100 ml beaker. Dip a piece of pH indicator paper in the water. Match the color of the wet pH paper to the color chart on the pH paper container to determine the pH of distilled water. Write down all of your results. You may use the lab notes section on the right.	
2.	Use the eyedropper to add 5 drops of vinegar to the distilled water and determine its pH.	
3.	Add another 5 drops of vinegar to the water and determine its pH.	
4.	Do this one more time and determine its pH.	
5.	Describe the trend in the pH values as more vinegar (acetic acid) is added. If you added more OH ions to water, the pH would increase.	
6.	Add about ½ teaspoon of sodium bicarbonate (baking soda, NaHCO ₃) to the water vinegar mixture. Describe what happens. Check the pH after the NaHCO ₃ is added. What did the NaHCO ₃ do to the pH?	
slig	me plants (especially grasses) grow best when the soil is htly acidic and dark green leafy plants and legumes (beans) or slightly basic soil.	
Digestive enzymes in your stomach function best in a strong acidic solution and the enzymes in the small intestine function best in a slightly basic fluid.		
	e following procedure demonstrates the hydrogen bonds ween water molecules.	
1.	Fill a drinking glass with a smooth rim with water. Use an eyedropper to add water until the water forms a slight elevated dome over the middle of the water surface. This dome shape of the water surface is called a meniscus.	
2.	Take a sewing needle and with a pair of tweezers gently lay the needle on the top of the meniscus so that it floats. This is possible because of the hydrogen bonds between the water molecules. It is as if the water molecules are holding hands underneath the needle holding it so that it floats.	
3.	With the eyedropper, add a drop of liquid soap to the water. What happens to the needle and the meniscus? The soap disrupts the hydrogen bonds between the water molecules.	

Biology

pH and Hydrogen Bonding of Water Molecules

Day 10

Lesson 2 Laboratory Report Name

Laboratory Report (20 points possible)

The pH of a solution

Hydrogen bonding between water molecules

Fill in the Blank

1.	Smaller units of larger molecules are called			
2.	The larger molecules that are composed of many of the smaller units are called			
3.	A 6 carbon carbohydrate molecule is called a			
4.	Two 6 carbon sugar molecules bonded together are called			
5.	Many 6 carbon sugar molecules bonded together are called			
6.	Draw or write the molecular formula for glucose.			
	Fructose is found in			
8.	Fructose is an of glucose, meaning that it has the same number of each as in glucose but they are in a different			
9.	Sucrose (table sugar) is formed by bonding together a molecule and a molecule.			
10.	Glucose is a molecule.			
11.	Sucrose is a molecule.			
12.	Maltose is formed by bonding a molecule to a molecule.			
13.	Lactose is formed by bonding a molecule to a molecule.			
14.	Lactose is found in			
15.	Many glucose molecules bonded together form a molecule.			
16.	This is an example of a			
17.	Two common starch molecules found in plants are and			
18.	Starch in animal tissues is called			
19.	Starch serves the purpose of in plant and animal cells.			
20.	Cellulose is a polymer formed by bonding of many molecules in such a way that grazers and humans cannot digest it.			
21.	In grazers and termites, cellulose is digested by			
22.	is a polysaccharide found in the shells of insects and crustaceans.			

23. Lipids include and	·
24. They are (meaning	water fearing).
25. In contrast to water, they arenegative ends.	_, meaning that they do not have positive and
26. Animal fats are called	<u></u> .
27. The main backbone structure of animal fats is a	molecule.
28 are attached to this	backbone.
29. Fatty acids with no double bonds between their car acids.	bon atoms are called fatty
30. Fatty acids with one double bond between their car fatty acids.	rbon atoms are called
31. Fatty acids with many double bonds between their fatty acids.	carbon atoms are called
32. Fats with many fatty acids with many double bond	s are at room temperature.
33. Fats with many fatty acids with no double bonds as	re at room temperature.
34. Excess energy is first stored in the body's adipose tissue.	in the liver and then in in

Carbohydrates

Day 14

Lesson 3 Laboratory Name

La	ab 3 — Carbohydrates	Lab Notes:	
RF	EQUIRED MATERIALS		
]]]]]]	Corn starch Sucrose (table sugar) Distilled water Stopper (from the supply kit) Iodine solution (from the supply kit) Eyedropper (from the supply kit) Test tubes (from the supply kit) Metric ruler (from the supply kit) Wax pencil (from the supply kit) Food items (potato, apple, banana, and corn kernel)		
IN	TRODUCTION		
Carbohydrates are essential to biological systems. Starches are important storage molecules that can be converted to monosaccharides yielding large amounts of useable energy.			
Glucose is a 6 carbon sugar with the formula $C_6H_{12}O_6$. Fructose (found in fruit) is also a 6 carbon sugar molecule with the same formula. These are monosaccharides (1 sugar unit).			
wh mo	To glucose molecules combine to form the sugar maltose, which is used in making malted milk shakes. A glucose blecule combined with a fructose molecule is sucrose (sugar un place in your food). These are disaccharides (2 sugar units) the formula $C_{12}H_{22}O_{11}$. Starches are long chains of glucose		
	olecules called polysaccharides (many sugar units).		
PURPOSE			
bro	is lab demonstrates how starches from food sources are oken down by saliva in the mouth before passing into the gestive tract.		
PF	ROCEDURE		
Answer the questions in this lab in your lab report with complete sentences.			
1.	Use the wax pencil to place a mark on 4 test tubes 3 cm		
•	(centimeters) up from the bottom. Number the test tubes from 1 to 4.		
2.	Add just enough corn starch to cover the bottom of each test tube marked #1 and #2. Add just enough sugar		
	(sucrose) to cover the bottom of each test tube marked #3 and #4.		

3.	To test tubes #1 and #3 add distilled water up to the 3 cm mark that you made. Place a stopper on top of each test tube and shake each well so that the starch and sugar dissolve.	
4.	Spit into test tubes #2 and #4 (yes, you read it right) and add distilled water up to the 3 cm mark. You have very special saliva (spit) that buffers the pH and adds the enzyme amylase that breaks down plant starch. Place a stopper on top of test tubes #2 and #4 and shake them	
	well.	
5.	Let all 4 test tubes sit for 30 minutes. If you see anything settle out of any test tube, shake it again.	
6.	After 30 minutes, add 1 drop of iodine (Lugol's) solution to each test tube.	
7.	Describe what you observe in each test tube. Test tube #1	
<i>/</i> •	has starch, water, and iodine. Test tube #3 has sucrose,	
	water, and iodine. These test tubes do not contain amylase,	
	so they are called the controls . Test tube #2 has starch, water, iodine, and amylase and test tube #4 has sucrose, water, iodine, and amylase. These are the experimental mixtures. Iodine turns a dark blue or black color with	
	starch but does not change color (react) with sucrose.	
8.	What is the effect of amylase on starch? What is the effect	
	of amylase on sucrose? Does amylase break down starch?	
	Does amylase break down sucrose?	
9.	Cut a thin slice of potato, apple, banana, and corn (cut the	
	kernel in half). Apply a drop of iodine to each. Which has a lot of starch? How can you tell?	
	a lot of states. From early ou ten.	

Laboratory Report (20 points possible)

- 1. Results from test tube 1
- 2. Results from test tube 2
- 3. Results from test tube 3
- 4. Results from test tube 4
- 5. The effect of amylase on starch
- 6. The effect of amylase on sucrose

Starch content of potato

Lab result

Conclusion and explanation

Starch content of apple

Lab result

Conclusion and explanation

Starch content of banana Lab result Conclusion and explanation

Starch content of corn Lab result Conclusion and explanation

Quizzes

Lesson 1 Quiz 1

Name

Match the Correct Answers

_____ 1. Electron A. protons and neutrons _____ 2. Mass of an atom B. 8 neutrons _____ 3. Atomic number C. same number of protons but not neutrons _____ 4. Isotope D. negative charge ____ 5. ¹⁴C₆ E. protons

Ci	rcle the correct answers	
6.	A glucose molecule has carbon atoms.	
	A. 2	
	B. 4	
	C. 6	
	D. 12	
7.	Metals tend to electrons.	
	A. gain	
	B. lose	
	C. annihilate	
8.	A proton has a charge of	
	A. +1	
	B1	
	C. 0	
9.	Metals and non-metals form	bonds.
	A. metallic	
	B. covalent	
	C. ionic	
	D. hydrogen	
10.	Non-metals and non-metals form	bonds
	A. metallic	
	B. covalent	
	C. ionic	
	D. hydrogen	

11.	is formed in photosynthesis.
	A. CO ₂
	3. O ₂
	$C. H_2O_2$
	$C_{12}H_{22}O_{11}$
12.	is not an ion.
-	A. O_2
	3. Cl ⁻
	C. Na ⁺
	D. H ⁺
13.	Elements are identified by their number of
-	A. protons
	3. neutrons
	C. electrons
	D. glyptons
14.	Schrodinger called the energy levels of electrons
-	A. layers
	3. orbits
	C. orbitals
	D. calories
15.	t was determined that atoms were more stable if they had electrons in their outer shell.
	A. 2
	3. 4
	C. 6
	D. 8

Exams

Match the Correct Answers

- _____ 1. Glucose
- _____ 1. Glucoso _____ 2. Metal
- _____ 3. Non-metal
- 4. Covalent bond
- _____ 5. Ionic bond
 - ____ 6. Isotope
- ____ 7. Atomic number
- ____ 8. More stable
- _____ 9. Water
- _____ 10. Hydrogen bonds
- _____ 11. pH 3
- _____ 12. pH 9
- ____ 13. ⁻OH
- ____ 14. Ice
- 15. Orbitals
- ____ 16. Sucrose
- _____ 17. Glycogen
- ____ 18. Triglycerides 19. Cellulose
- _____ 1). Genurose
- _____ 20. Saturated fatty acid
- _____ 21. Starch
- ____ 22. Adipose tissue
- _____ 23. Protein
- 24. –COOH
- _____ 25. Uracil

- A. metal and non-metal
- B. 6 carbon atoms
- C. gains electrons
- D. loses electrons
- E. non-metal and non-metal
- A. 8 electrons in outer shell
- B. + and poles
- C. same number of protons, different number of neutrons
- D. raise boiling point of water
- E. protons
- A. base
- B. less dense than water
- C. Schrodinger
- D. more base than acid
- E. more acid than base
- A. animal starch
- B. no double bonds
- C. polysaccharide
- D. glucose, fructose
- E. animal fat
- A. stores lipids
- B. carboxyl
- C. RNA
- D. stored in liver
- E. chain of amino acids

 26. Ribose	A.	backbone of nucleic acid
 27. Thymine	B.	tertiary structure of protein
 28. Sugar-phosphate	C.	RNA
 29. ATP	D.	high energy molecule
30. Folded spiral helix	E.	DNA

Answers to Worksheets, Laboratory Reports, Quizzes and Exams

Biology — Worksheet and Lab Report Answer Keys

Lesson 1 Worksheet

- 1. Element
- 2. Protons, neutrons, electrons (in any order)
- 3. Protons
- 4. Element
- 5. Molecules
- 6. $12 \text{ H}_2\text{O} + 6 \text{ CO}_2 \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6 \text{ O}_2 + 6 \text{ H}_2\text{O}$
- 7. 6
- 8. 6
- 9. Glucose See Figure 1.1
- 10.12
- 11.6
- 12. Protons, neutrons (any order)
- 13. +1
- 14. Zero
- 15. -1
- 16. Protons, neutrons (any order)
- 17. Protons, neutrons (any order)
- 18. Protons, neutrons (in this order)
- 19.7,7
- 20. Yes. They both have 6 protons and they have different numbers of neutrons.
- 21. No. They have different numbers of protons.
- 22. Orbitals
- 23. 2
- 24. 8
- 25. Lose
- 26. Gain
- 27. Ionic
- 28. Attraction
- 29. Positive, negative (in this order)
- 30. Share, covalent
- 31. 2, 4, 4, 8
- 32, 4

Lesson 1 Lab Report

The student is to go to a place outdoors and describe the observed life forms. The report is to consist of complete sentences and clear descriptions. Award a possible 20 points for the overall descriptions:

- 8 points for identifying and describing a keystone species (one that if removed would have a large impact on the others)
- 6 points for identifying and describing consumers (feeders)
- 6 points for identifying and describing decomposers. The impacts of bacterial decomposition will be seen rather than the bacteria themselves. Mold also counts as a decomposer.

Lesson 2 Worksheet

- 1. Negative, positive (in this order)
- 2. Attracted
- 3. hydrogen
- 4. Negative, positive (In this order)
- 5. Non-polar
- 6. Hydrophilic, hydrophobic
- 7. 1 gram
- 8. High, a lot of
- 9. Less, floats
- 10. Hydrogen, -200
- 11. Hydrogen
- 12. 10,000,000 or 10 million or 10⁷, H⁺ and ⁻OH
- 13. H⁺
- 14. -OH
- 15. Neutral
- 16. 100, OH or base
- 17. 10, H+ or acid

Lesson 2

Lab Report

When grading the lab report, award

- 10 possible points for following directions
- 10 points for answering the questions for the pH procedures and hydrogen bonding procedures

The answers are the student's observations. 20 points possible overall.

Lesson 3 Worksheet

- 1. Monomers
- 2. Polymers
- 3. Monosaccharide
- 4. Disaccharides
- 5. Polysaccharides
- 6. $C_6H_{12}O_6$
- 7. Fruit
- 8. Isomer, atom, arrangement
- 9. Glucose, fructose (either order)
- 10. Monosaccharide
- 11. Disaccharide
- 12. Glucose, glucose
- 13. Glucose, galactose (either order)
- 14. Milk
- 15. Starch
- 16. Polysaccharide
- 17. Amylose, amylopectin (either order)
- 18. Glycogen
- 19. Energy storage
- 20. Glucose
- 21. Microorganisms
- 22. Chitin
- 23. Fats, oils (either order)
- 24. Hydrophobic
- 25. Non-polar
- 26. Triglycerides
- 27. Glycerol

- 28. Fatty acids
- 29. Saturated
- 30. Monounsaturated
- 31. Polyunsaturated
- 32. Liquids (oils)
- 33. Solids
- 34. Glycogen, fat

Lesson 3 Lab Report

The report for steps 1–8 describes the results for test tubes 1–4. The answers to the questions are the student's observations. Award

- 15 possible points for steps 1–8
- 5 possible points for the answers (student observations) for step 9. Iodine turns food dark blue or black when starch is present

20 points possible overall.

Lesson 4 Worksheet

- 1. Muscle tissue, cell membranes, and enzymes (any order)
- 2. Amino acids
- 3. Amino, carboxyl (in this order)
- 4. Acid
- 5. R
- 6. Glycine, H (hydrogen)
- 7. Peptide
- 8. Dipeptide
- 9. Polypeptide
- 10. Sequence of amino acids
- 11. Spiral helix or sheet formed by the chain of amino acids
- 12. Spiral helix or sheet folding upon itself
- 13. Attachment of two or more polypeptide chains to each other
- 14. Enzymes
- 15. Substrate
- 16. DNA, RNA

Biology — Quiz Answer Keys

Quiz #1 Chemical Principles in Biology

- 1. D
- 2. A
- 3. E
- 4. C
- 5. B
- 6. C
- 7. B
- 8. A
- 9. C
- 10. B
- 11. B
- 12. A
- 13. A
- 14. C
- 15. D

Quiz #2

Water

- 1. D
- 2. B
- 3. B
- 4. D
- 5. C
- 6. A
- 7. C
- 8. B
- 9. D
- 10. A
- 11. B
- 12. C
- 13. D
- 14. A
- 15. E

Quiz #3

Carbohydrates and Lipids

- 1. E
- 2. A
- 3. D
- 4. C
- 5. B
- 6. D
- 7. B
- 8. C
- 9. C
- ,. –
- 10. B
- 11. B
- 12. D
- 13. C
- 14. E
- 15. A

Quiz #4

Proteins and Nucleic Acids

- 1. A
- 2. D
- 3. B
- 4. E
- 5. C
- 6. B
- 7. A
- 8. C
- 9. B
- 10. D
- 11. E
- 12. A
- 13. C
- 14. D
- 15. B

Biology — Exam Answer Keys

Examination #1 (Lessons 1-5)

- 1. B
- 2. D
- 3. C 4. E
- 5. A
- 6. C
- 7. E
- 8. A
- 9. B
- 10. D
- 11. E
- 12. D
- 13. A
- 14. B
- 15. C
- 16. D
- 17. A
- 18. E
- 19. C
- 20. B
- 21. D 22. A
- 23. E
- 24. B
- 25. C
- 26. C
- 27. E
- 28. A
- 29. D
- 30. B

Examination #2 (Lessons 5-8)

- 1. E
- 2. C
- 3. A

- 4. D
- 5. B
- 6. D
- 7. C
- 8. E
- 9. B
- 10. A
- 11. B
- 12. A
- 13. D
- 14. E
- 15. C
- 16. C
- 17. D
- 18. A

- 19. B
- 20. E 21. C
- 22. B
- 23. E
- 24. D
- 25. A
- 26. B
- 27. D
- 28. A
- 29. E
- 30. C

Examination #3 (Lessons 9-12)

- 1. B
- 2. D
- 3. E
- 4. C
- 5. A
- 6. D
- 7. A